**EXERCISE #10: counting π electrons and determining the presence of aromaticity**

Organic Chem II Alfred State College

**10a) Determine the # of pπ electrons in each of the molecules below**

*notes on pπ e-  counting*

...2pπ electrons for each double bond; radicals may count as pπ as long as all sigma bonds are first satisfied and filled;







CH2=CH2

(C2 H4) (C3H4) (C3H3\*radical) C3H3 + (cation)

=# pπ

**AROMATIC ?**

yes no yes no yes no yes no









= # pπ

**AROMATIC ?**

yes no yes no yes no yes no

**b) determine whether each of the molecules above is aromatic** (circle yes no )

1) must be flat

2) must have conjugation (at least one double-single combo)

3) must obey 4n+2 pπ count rule