HOMEWORK ASSIGNMENT #5 ORGANIC CHEMISTRY I (20 pts)

**free radical reactions & higher alkanes**

 **(due Monday 12 October)**

 **Your name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_**

**5.1a.** In early experiments in the l920s, Fritz Paneth and co-workers at the University of Berlin passed

tetramethyl lead, (CH3)4Pb, down an evacuated glass tube. They heated the tube with a coil of hot wire wrapped around the outside of the glass. Where ever the coil was wrapped, a shiny deposit of pure Pb metal appeared nearby and a stream of ethane (CH3-CH3) was detected at the exit end of the tube as sketched below. (3 pts)

 **Hot wire**

**Pb(CH3)4 in** **CH3-CH3 out**

  **Pb metal deposit forms**

Suggest a mechanism for this set of observations.

5.1. If the hot wire is moved up stream closer to the original source of the Pb(CH3)4, and more Pb(CH3)4 is flowed

 into the tube the original Pb metal deposit disappears, a new one forms were the wire is placed, and, mostly

 just Pb(CH3)4 exits the tube.

 **New Pb deposit forms old Pb deposit slowly disappears**

**Pb(CH3)4 in**

 **Pb(CH3)4 out Hot wire**

 **moved upstream**

**Explain these results in terms of the mechanism in 5.1a you propose. (3 pts)**

**5.2 Drawn and quartered… 3 pts**

 Sketch on the same plot below the reaction coordinate diagrams of Br and Cl with CH4 , making sure to emphasize where they differ.

 Energy

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 **CH4 + Br or Cl**

 **reaction progress**

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**5.4. Circle the carbon at which the most rapid chlorination is expected**

 **in each of the two structures below ( 2 pts)**



**5.5) Productive thinking (4 pts)**

Photochemical bromination of 2,2,4-trimethylpentane yields 4 different monobromides. Draw them below

A (1o) B (1o)

C(2o) D (3o)

6.1b Compute the expected % yields of A-D you’ve drawn above given the following relative yield (R) data below: (see also- exercise 7 and supplement 7). 5 pts

 **Site degree** **Relative yield/H for radical Bromination**

 1o  1

 20` 50

 30 750

**Compound expected % yield**

A \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

B \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

C \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

D \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

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