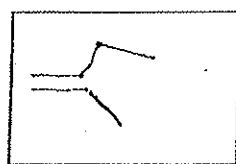
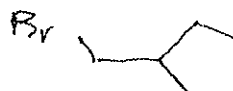
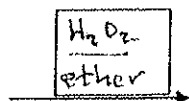
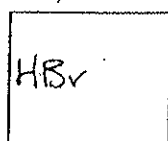


exercise #18 boxing lessons with alkenes
carbocation, halohydrin and radical additions and alkene syntheses
Organic Chem I Alfred State College

1)



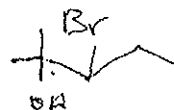
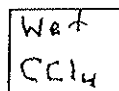
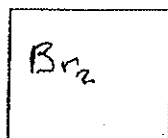
+



1-bromo-2-methylbutane

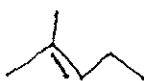
2)

2-methyl-2-pentene +



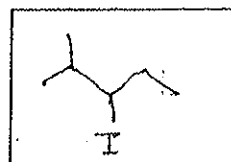
3-bromo-2-methyl-2-pentanol

3)



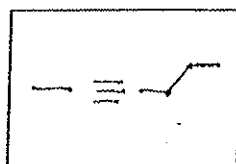
2-methyl-2-pentene + HI

$\xrightarrow{\text{Peroxide/acetic acid}}$

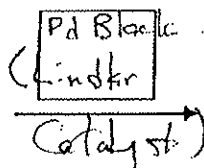
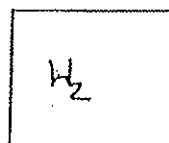


ant: Mark.
add.

4)



+

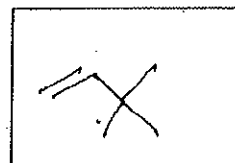
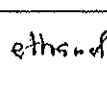


Z-only 2-hexene



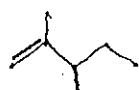
5)

3,3-dimethyl-2-bromobutane + OH^-

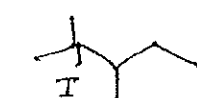
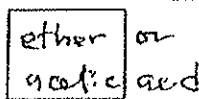
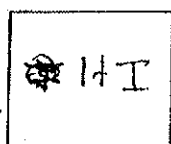


only

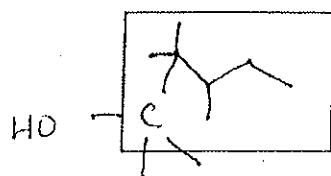
6)



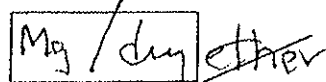
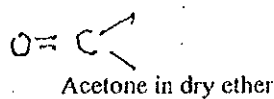
2,3-dimethyl-1-pentene +



2,3-dimethyl-2-iodopentane

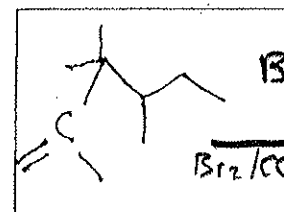


$\xrightarrow{\text{H}^+/\text{H}_2\text{O}}$

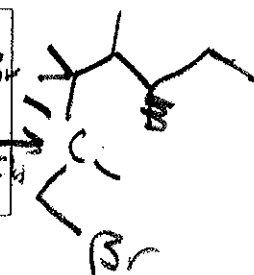


Phosphoric
acid/reflux

Br_2/CCl_4

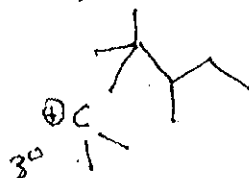


dehydration



rearrange?

if no rearrangement



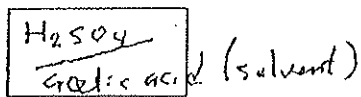
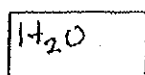
$\rightarrow$

no higher degree carbocation
So no rearrangement

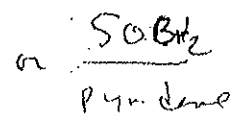
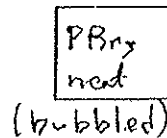
exercise 18(continued)

7)

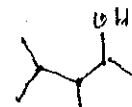
2-methylpropene +



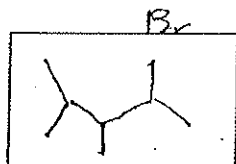
2-methyl-2-propanol



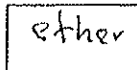
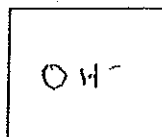
2-methyl-2-bromopropane



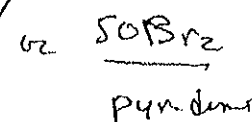
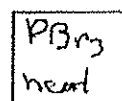
8)



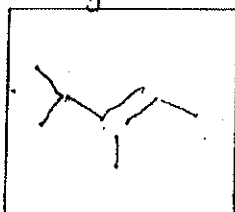
+



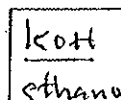
3,4-dimethyl-2-pentanol



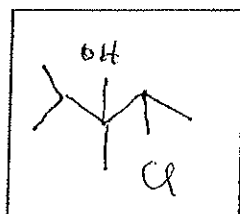
major



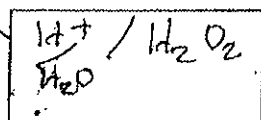
+ 3,4-dimethyl-1-pentene
(minor)



Cl_2 in wet CCl_4

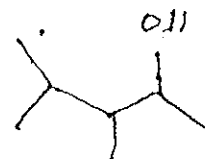


Conc
sulfuric
acid



ant. add. of $H-O-H$

3,4-dimethyl-2-pentanol
(where it all began...)



then H_2O

