1. Define Bronsted Acids and Bases and differentiate from Lewis acids and bases.
2. Explain why the alcohol is the stronger acid. Use the conjugate base in your reasoning.



1. What are the four things to consider when predicting acidity? Provide an example for each and explain.
2. For the following reactions,
	1. determine which acid is more acidic and explain why
	2. draw a mechanism (curved arrows), and
	3. predict whether the reaction favor the right or left under equilibrium conditions







