**Mini-quiz #34 Chemistry 1114 section 2 (Fong) 1 Dec 2014 4 pts A**

**Your name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_**

**Use the Periodic Table displayed on the projection screen in class to determine the correct formula for ionic compounds prepared by combining the elements below:**

**a) Ca and P Ca3P2**

**b) Li and O Li2O**

**c) B and F BF3**

**d) Al and O Al2O3**

**Mini-quiz #34 Chemistry 1114 section 2 (Fong) 1 Dec 2014 4 pts B**

**Your name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_**

**Use the Periodic Table displayed on the projection screen in class to determine the correct formula for ionic compounds prepared by combining the elements below:**

**a) Mg and N Mg3N2**

**b) Na and Se Na2Se**

**c) B and O B2O3**

**d) Al and Cl AlCl3**

**Mini-quiz #34 Chemistry 1114 section 2 (Fong) 1 Dec 2014 4 pts C**

**Your name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_**

**Use the Periodic Table displayed on the projection screen in class to determine the correct formula for ionic compounds prepared by combining the elements below:**

**a) Mg and F MgF2**

**b) Ca and Se CaSe**

**c) Al and S Al2Se3**

**d) K and C K4C**