**Mini-quiz #19 Chemistry 1114 Wednesday 6 November 2013**

**5 pts**

**Your name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ A**

Use the Periodic Table on the flip side of this page to determine the likely stable formulas for the ionic and near-ionic binary combinations of the elements paired below

**Element Most likely**

**Pair formula**

K + Se

Mg + As

Li + N

Rb + Br

Al + S

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**5 pts**

**Your name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ B**

Use the Periodic Table on the flip side of this page to determine the likely stable formulas for the ionic and near-ionic binary combinations of the elements paired below

**Element Most likely**

**Pair formula**

Na + Te

Ca + N

B + P

Cs + F

Ga + As