**Chem 1013: mini-quiz # 20: molecular, complete ionic, net ionic A**

**Your name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_**

**Write the complete molecular reaction for the compounds below assuming that AgCl is the only species formed as a final precipitate**

**AgNO3(aq) + KCl(aq)-🡪**

**Write the equivalent complete ionic reaction for the reaction above: (note- Ag is +1**

**NO3 is -1, K is +1 and Cl is -1)**

**Write the equivalent net ionic reaction for the reaction above:**

**Chem 1013: mini-quiz # 20: molecular, complete ionic, net ionic B**

**Your name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_**

**Write the complete molecular reaction for the compounds below, assuming that AgCl is the is the only species formed as a final precipitate**

**AgClO3(aq) + LiCl(aq)-🡪**

**Write the equivalent complete ionic reaction for the reaction above: (note- Ag is +1**

**ClO3- is -1, Li is +1 and Cl is -1)**

**Write the equivalent net ionic reaction for the reaction above:**